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Silicon-Directed Oxa-Pictet—Spengler Cyclization and an Unusual Dimerization of 2-Trimethylsilanyl Tryptophols

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ABSTRACT

The tetrahydro-pyrano[3,4-b]indoles 6 were synthesized from 2-(2-trimethylsilanyl-1*H*-indol-3-yl)-ethanols 5 and various ketones or aldehydes through silicon-directed oxa-Pictet—Spengler cyclizations. An unusual reaction led to the dimeric products 7 when some of 5 was treated with acetone using BF₃ as the catalyst.

The tetrahydro-pyrano[3,4-*b*]indole scaffold **1** (Figure 1) has been widely explored in the field of pharmaceutical research. One of the more successful examples, etodolac (**2**) (brand name: Lodine), was discovered as a novel cyclooxygenase-2 (COX-2) inhibitor.¹ This clinical success, along with recent reports that COX-2 may play an important role during premalignant hyperproliferation,² has triggered the reinvestigation of tetrahydro-pyrano[3,4-*b*]indoles in the areas of both synthetic and medicinal chemistry.

Versatile methods for synthesizing tetrahydro-pyrano[3,4-*b*]indoles have been documented in the literature. The most efficient method is the traditional oxa-Pictet—Spengler acid-catalyzed cyclocondensation of tryptophols with either ketones or aldehydes.^{3,4} However, this method appears to be limited to the assembly of electron-rich or electron-neutral

$$R_1$$
 R_2
 R_3
 R_4
 R_3
 R_4
 R_4
 R_4
 R_5
 R_4
 R_4
 R_5
 R_5
 R_6
 R_7
 R_8
 R_8

Figure 1.

tetrahydro-pyrano[3,4-*b*]indoles. Moreover, the tryptophol precursors for the oxa-Pictet—Spengler cyclization are generally prepared through classical methods such as Fisher, Madelung, and Reissert procedures,⁵ which are limited both in the degree of substitution and the type of functionality that can be incorporated into the indole nucleus. During the course of structure—activity relationship studies, we needed

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Scheme 1

R₁

$$R_1$$
 R_2
 $NIS, THF: MeOH$
 R_2
 NH_2
 NH_2
 R_1
 NH_2
 NH_2
 R_1
 NH_2
 R_1
 R_2
 NH_2
 R_1
 R_2
 NH_2
 R_1
 R_2
 R_2
 R_3
 R_4
 R_4
 R_4
 R_4
 R_4
 R_4
 R_4
 R_5
 R_6
 R_6
 R_7
 R_8
 R

to prepare a series of electron-withdrawing group-substituted tetrahydro-pyrano[3,4-*b*]indoles. Here, we describe an efficient methodology for syntheses of various electron-demanding tetrahydro-pyrano[3,4-*b*]indoles using Larock's palladium-catalyzed indole synthesis followed by silicon-directed oxa-Pictet—Spengler cyclization of resultant tryptophols with ketones or aldehydes. The scope and limitation of this methodology are also discussed.

The required tryptophol components, 2-(2-trimethylsilanyl-1*H*-indol-3-yl)-ethanols, and 2-(2-trimethylsilanyl-1*H*-indol-3-yl)-propanols **5** were easily prepared in two steps as described by Larock (Scheme 1).⁶ First, selective ortho iodinations of anilines **3** afforded the 2-iodinated products **4** in modest to good yields.⁷ Palladium-catalyzed heteroannulation of *ortho*-iodoanilines with 4-(trimethylsilanyl)-3-butyn-1-ol or 5-(trimethylsilanyl)-4-pentyn-1-ol gave the corresponding 2-(2-trimethylsilanyl-1*H*-indol-3-yl)-ethanols or 2-(2-trimethylsilanyl-1*H*-indol-3-yl)-propanols **5** with no regioisomeric products isolated. Generally, this reaction proceeded in reasonable yields, excellent regioselectivity, and good tolerance for substitution on the indole cores.

Our initial study on the oxa-Pictet—Spengler cyclization started with 2-(2-trimethylsilanyl-1*H*-indol-3-yl)-ethanol **5a** (entries 1 and 2). As expected on the basis of the well documented behavior of tryptophol, the tetrahydro-pyrano-[3,4-*b*]indole **6a** was obtained in reasonable yield when **5a** was condensed with acetone in trifluoroacetic acid (TFA). To our surprise, when BF₃ etherate was employed as the promoter the reaction of **5a** with acetone always produced

complicated mixtures. We tried optimizing the reaction conditions by varying time, temperature, and concentration. Ultimately, a dimeric species was isolated in 45% yield, and the structure was determined to be 7a. To reconcile the divergent results from TFA- and BF₃-promoted cyclization experiments, we prepared the substrates 5 containing various substitution patterns on the indole cores. These results are summarized in Table 1.8 Generally, regardless of the strength of the electron-withdrawing groups on the silvl tryptophols 5, condensation with acetone under TFA promotion to produce tetrahydro-pyrano[3,4-b]indoles 6 proceeded in moderate to good yields without detection of the dimers 7 (entries 1, 7, 10, 12, 14, and 18). Benzophenone under these reaction conditions failed to give any cyclization adduct probably due to the steric hindrance (entry 13). Instead, the desilylated product 8a was obtained after workup. Similarly, the uncyclized adducts 9a and 9b were obtained when the electron-deficient trytophols 5g and 5h were employed for the condensations with cyclohexanone under identical conditions (entries 15 and 19). It is worth noting that 10 could be generated from the reaction of homotryptophol 5i with CH₃-CHO in satisfactory yield (entry 21, 54%), though the similar reaction with acetone only gave the desilylated product 8b without any cyclization (entry 20).

A possible mechanism for the silicon-promoted formation of the tetrahydro-pyrano[3,4-b]indoles **6** is shown in Scheme 2. Two pathways are likely involved in this process. Pathway A involves the acid-catalyzed formation of the oxocarbenium ion species **12**, followed by intramolecular electrophilic aromatic substitution on the activated position *ipso* to the trimethylsilyl group of **12**. Intermediate **13** is then believed to form due to the stabilizing hyperconjugation of the TMS group on the β -carbocation. Desilylation of **13** then affords

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 $^{(7) \} Iodo-4, 5-dichloroaniline \ and \ 2-iodo-4-fluoro-5-chloroaniline \ were obtained along with their regioisomers.$

^{(8) (}a) Typical Procedure for TFA-Promoted Oxa-Pictet-Spengler Cyclization, Preparation of 6i. Acetone (0.2 mL) was added into 5f (100 mg, 0.35 mmoL) in 3 mL of 2:1 CH₂Cl₂/trifluoroacetic acid solution at 0 °C. The reaction was stirred for 2 h from 0 °C to rt. The reaction was then quenched with saturated NaHCO3 solution and extracted 3x with CH2Cl2. The organic layer was washed with brine, dried over anhydrous Na₂SO₄, filtered, concentrated, and purified by column chromatography (silica gel, hexanes/ EtOAc from 2:1 as the eluent) to give 6i as a white solid (75 mg, 81%): ¹H NMR (CDCl₃) δ 7.60 (br, s, 1H), 7.16 (d, J = 3.5 Hz, 1H), 7.04 (d, J = 8.5 Hz, 1H), 3.92 (t, J = 8.5 Hz, 2H), 2.62 (t, J = 8.5 Hz, 2H), 1.42 (s, 6H); 13 C NMR (CDCl₃) δ 154.2, 151.8, 141.2, 131.8, 114.7, 112.3, 106.9, 104.4, 71.6, 60.2, 27.6, 22.1; MS (m/z) MH⁻ 284. Anal. Calcd for C₁₃H₁₃CIFNO: C, 61.54; H, 5.16. Found: C, 61.23; H, 5.09. (b) **Typical** Procedure for BF₃-Promoted Dimerization, Preparation of 7c. BF₃ etherate (0.23 mL, 3.70 mmoL) was added dropwise into the mixture of 5f (480 mg, 1.68 mmoL) and acetone (0.14 mL, 1.85 mmoL) in 5 mL of dichloromethane at 0 °C. The reaction was slowly warmed to rt over 2 h, and stirring was continued for another 4 h. The mixture was then poured into saturated NaHCO3 solution and extracted three times with CH2Cl2. The organic layer was washed with brine, dried over anhydrous Na₂SO₄, filtered, concentrated, and purified by column chromatography (silica gel, hexanes/EtOAc from 3:1 to 1:1 as the eluent) to give the dimer 7c as a white solid (285 mg, 67%): ${}^{1}H$ NMR (CDCl₃) δ 7.66 (br, s, 1H), 7.28 (d, J = 9.0 Hz, 1H), 7.24 (d, J = 4.0 Hz, 1H), 7.20 (d, J = 8.5 Hz, 1H), 7.01 (d, J = 4.5 Hz, 1H), 3.88 (m, 2H), 3.76 (m, 1H), 3.52 (m, 1H), 3.05 (t, J = 8.5 Hz, 2H), 2.98 (abq, J = 13.5 Hz, 1H), 2.85 (m, 2H), 2.62 (abq, J = 13.5 Hz, 1H), 2.85 (m, 2H), 2.85 (m, 2 13.5 Hz, 1H), 2.11 (s, 3H), 1.53 (s, 3H), 1.31 (s, 3H); ¹³C NMR (CDCl₃) δ 152.9, 150.6, 149.8, 139.4, 131.4, 129.2, 127.3, 125.8, 115.6, 114.3, 113.3, 112.3, 111.0, 109.8, 109.0, 106.0, 104.2, 103.3, 101.1, 61.6, 60.8, 56.0, 37.0, 28.4, 27.5, 26.4, 26.0; IR (film, cm⁻¹) 3362, 3290, 1470; MS (*m/z*) MH⁺ 507, MNa⁺ 529. HRMS calcd for C₂₆H₂₆F₂N₂O₂Cl₂ 506.1339, found 506.1342.

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Table 1. Reaction of Silanyl Derivatives **5** with Carbonyls under CF₃COOH or BF₃ Etherate Promotion^{*a,b*}

| entry | substrate 5 | promotor | reagent | product 6 or 7 and |
|--------|-------------------------------|----------------------|---------------------------------------|--------------------|
| | | | (ketone or aldehyde) | isolated yield (%) |
| 1 | $R_1 = H, R_2 = H$ | CF ₃ COOH | CH3COCH3 | 6a (51%) |
| 2 | 5a | BF_3 | CH3COCH3 | 7a° (45%) |
| | | | | ` ′ |
| 3 | $R_1 = CN, R_2 = CF_3$ | BF_3 | CH3COCH3 | 6b (95%) |
| 4 | 5b | | CH3COCH2SMe | 6c (68%) |
| 5 6 | | | CH3COCH2SEt | 6d (65%) |
| 6 | | | CH ₃ COCH ₂ SBn | 6e (62%) |
| 7 | $R1 = C1, R_2 = CF_3 5c$ | CF3COOH | CH3COCH3 | 6f (75%) |
| 8 | $R1 = NO_2$, $R_2 = CF_3$ 5d | \mathbf{BF}_3 | CH3COCH3 | 6g (60%) |
| 9 | $R_1 = Cl, R_2 = Cl 5e$ | BF_3 | CH ₃ COCH ₃ | 7b (80%) |
| 10 | | CF3COOH | CH ₃ COCH ₃ | 6h (90%) |
| 11 | $R_1 = F_1 R_2 = C1$ | BF_3 | CH ₃ COCH ₃ | 7c (67%) |
| 12 | 5f | CF ₃ COOH | CH₃COCH₃ | 6i (81%) |
| 13 | | | PhCOPh | F_OH |
| | | | | CI N |
| | | | | H 8- (70%) |
| | | | | 8a (70%) |
| 14 | $R_1 = CF_3, R_2 = H$ | CF ₃ COOH | CH₃COCH₃ | 6j (50%) |
| 15 | | Cr ₃ COOn | Cyclohexanone | F ₅ C |
| 15 | 5g | | Cyclonexanone | UN OH |
| | | | | "() |
| | | | | 9a (65%) |
| 16 | $R_1 = CO_2Me$, $R_2 = H$ | BF_3 | CH ₃ COCH ₃ | 7d (80%) |
| 17 | 5h | | CH3COCH2SEt | 6k (20%) |
| 18 | | CF ₃ COOH | CH ₃ COCH ₃ | 6l (82%) |
| 19 | | | Cyclohexanone | Mecocc |
| | | | | |
| | | | | 6m (30%) + |
| | NG A A A | om oo ou | OTT 00 0TT | 9b (60%) |
| 20 | F-C N TMS | CF₃COOH | CH₃COCH₃ | ELLI OH |
| | 5i | | | PL (920/) |
| | 31 | | | 8b (83%) |
| 21 | | | CH3CHO | NC |
| 21 | | | Спзспо | F,C N O |
| | | | | 10 (58%) |
| | | | | |

 a CF₃COOH procedure: **5** (1.0 equiv), ketone or aldehyde (1.5~2.0 equiv) in CF₃COOH (c=0.1 M) at 0 °C, then warm to rt, continue for 4~8 h. b BF₃ etherate procedure: **5** (1.0 equiv), ketone (1.5~2.0 equiv), BF₃ etherate (2.0~3.0 equiv) in CH₂Cl₂ (c=0.1 M) at 0 °C, then warmed to rt, continue for 4~8 h. c Complex mixture was obtained, and **7a** was the only product that could be characterized. Replacing **5a** with trytophol gave a similar result

Scheme 2

Pathway A

$$R_1$$
 R_2
 R_3
 R_4
 R_4
 R_5
 R_4
 R_5
 R_4
 R_5
 R_5
 R_4
 R_5
 R_5
 R_5
 R_5
 R_5
 R_4
 R_5
 R

the tetrahydro-pyrano[3,4-b]indole **6**. In pathway B, a type of Mukaiyama adol condensation of **5** occurs also assisted

by TMS group hyperconjugation. Thus, desilylation of 11 forms the diol 14. S_N1-type of displacement of H₂O catalyzed by acid in 14 affords the tetrahydro-pyrano[3,4-b]indole 6. Dehydration of 14 was observed to give the isolated products **9a** and **9b** when cyclohexanone was employed as a ketone component. For less reactive ketones such as benzophenone, a simple protodesilylation process occurs to generate 8a. Overall, a TMS group at the α -position of indole 5, acting as a "big" hydrogen, could facilitate either the electrophilic aromatic substitution or Mukaiyama adol process and results in substitution by either an intermolecular or intramolecular electrophile. It is of interest that the proposed mechanism here has precedent in the literature described by Kohno and Sekine. 10 This modification on the precursor, compared with that of traditional oxa-Pictet-Spengler cyclization, is extremely effective when an electron-deficient tryptophol is employed as the substrate. For example, acid-catalyzed cyclization of the nonsilylated substrate of 5d (structure not shown) with acetone only gave a trace amount of 6g, with most starting material being recovered, while when more harsh conditions (strong acid and heat) were applied, decomposition of the substrate occurred.

The outcome of the BF₃-promoted cyclization appears to be related to the structures of both the tryptophols 5 and the ketones. Formations of the dimers 7a-d could be observed when acetone and electron-rich, electron-neutral, or weakly electron-deficient 2-(2-trimethyl-silanyl-1*H*-indol-3-yl)-ethanols such as 5a, 5e, 5f, and 5h were employed in the reactions (entries 2, 9, 11, and 16). BF₃-promoted reactions of 5 with the ketones other than acetone afforded the tetrahydro-pyrano[3,4-b]indoles **6** in moderate to good yields (entries 4-6, 17, and 19). Surprisingly, in the case of the BF₃-promoted reactions of some strongly electron-deficient 2-(2-trimethylsilanyl-1*H*-indol-3-yl)-ethanols such as **5b** or 5d and acetone, the cyclization product formed was the tetrahydro-pyrano[3,4-b]indole **6b** or **6g**, respectively (entries 3 and 8). The structure and the relative stereochemistry of 7a were confirmed by X-ray crystallography. Formation of the dimer 7 appears to be due to the reaction between 5 and 4-hydroxy-4-methyl-2-pentanone 15 generated from BF₃catalyzed self-condensation of acetone. Supporting evidence for this pathway is illustrated in Scheme 4; utilizing 15 as the ketone component in the oxa-Pictet-Spengler process led to isolation of the dimer 7c. Although mechanistically unclear, this dimerization process can be envisioned as shown in Scheme 3. Acid-catalyzed dimerization of acetone is in equilibrium with dissociation of 15. It is more likely that the initial process is the same as pathway B shown in Scheme 2. Thus, the Mukaiyama adol condensation of **5** with **15** gave the intermediate 16, which acts as the electrophile at the gemdimethyl alcohol site for a Friedel-Craft-type process at the α -position of the second equivalent of 5 to form a dimeric species 17. It is probably due to the steric hindrance of the allylic alcohol site next to the indole core (the other tertiary alcohol) that the intramolecular process occurs only at the less hindered gem-dimethyl alcohol position. Finally, an intramolecular cyclization of 17 results in the dimer 7. In

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Scheme 3

Scheme 3

Scheme 3

R₁

$$R_1$$
 R_2
 R_1
 R_2
 R_1
 R_2
 R_3
 R_4
 R_4
 R_5
 R_5
 R_5
 R_7
 R_1
 R_7
 R_8
 R_8

the TFA-promoted reaction, the equilibrium shown in Scheme 3 favors the acetone side, leading to the normal oxa-Pictet—Spengler cyclization to form **6**. In comparison, using

BF₃ as the promoter in the reaction favors the 15 side probably due to the formation of the stable complex ion between BF₃ and the carbonyl group of 15.11 Thus, the dimer 7 is generated through the sequential steps shown in Scheme 3. Probably due to the strong electron-deficiency in **5b** and 5d, the poor nucleophilicity of silyl indole prohibits the Mukaiyama adol process. Instead, pathway A in Scheme 2 occurred to generate pyrano indoles 6b and 6g without any dimers 7. To our surprise, applying d_6 -acetone in BF₃promoted cyclization with 5f only gave the deuterated Pictet—Spengler product 18 in 91% yield (Scheme 4). It is reported that the primary isotope effect $(k_{\rm H}/k_{\rm D})$ of acidcatalyzed enolization of acetone, which is the rate-determining step of acetone self-condensation process, is about 7.¹² The substantial difference results in completely different reaction pathways between acetone and d_6 -acetone since the dimerization process to 7c is blocked by slow conversion of d_6 -acetone to d_{12} -4-hydroxy-4-methyl-2-pentanone. Unlike 15, its fully deuterated agent, d_{12} -4-hydroxy-4-methyl-2pentanone, when employed in BF₃-catalyzed reaction with **5f**, afforded **19** in 75% yield without detection of any deuterated dimer 7c (Scheme 4). The mechanism is still under investigation. It may be related to the isotope effect during the dissociation of the tert-OH to form the five-membered ring of the dimer 7.

In summary, we have successfully established an efficient synthetic method for the preparation of tetrahydro-pyrano-[3,4-*b*]indoles **6** by a Larock heteroannulation and silicondirected oxa-Pictet—Spengler cyclization process. From a synthetic point of view, this new procedure constitutes a significant improvement of the traditional oxa-Pictet—Spengler reaction by overcoming its limitation to electronrich or electron-neutral pyranoindoles. Finally, an unusual dimeric structure, **7**, was isolated and characterized during the BF₃ etherate-promoted reaction of 2-(2-trimethylsilanyl-1*H*-indol-3-yl)-ethanols **5** with acetone.

Supporting Information Available: Experimental procedures and full characterization of new compounds 5–7, 9, 10, 18, and 19. This material is available free of charge via the Internet at http://pubs.acs.org.

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